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(21) International Application Number: PCT/US99/24703 (22) International Filing Date: 21 October 1999 (21.10.99) (30) Priority Data: 09/184,196 2 November 1998 (02.11.98) US (71) Applicant (for all designated States except US): BASF CORPORATION [US/US]; 26701 Telegraph Road, Southfield, MI 48034-2442 (US). (72) Inventors; and (75) Inventors/Applicants (for US only): OHRBOM, Walter [US/US]; BASF Corporation, 13938 Plover, Hartland Township, MI 48353 (US). MENOVCIK, Gregory [US/US]; 26238 Farmington Road, Farmington Hills, MI 48334 (US). ST. AUBIN, Donald [US/US]; 505 Sherbrooke, Commerce Township, MI 48382 (US). BOISSEAU, John [US/US]; 2071 Birchland, Bloomfield Hills, MI 48302 (US). REHFUSS, John [US/US]; 15130 Chilgrove Lane, Huntersville, NC 28078 (US). (74) Agent: GOLOTA, Mary; BASF Corporation, 26701 Telegraph Road, Southfield, MI 48034-2442 (US).		(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: CURABLE COATING COMPOSITIONS CONTAINING BLENDS OF CARBAMATE-FUNCTIONAL COMPOUNDS (57) Abstract <p>The present invention provides a curable coating composition that includes at least three components. The coating composition includes a component (a) that includes one or both of a compound (a)(1) having at least one carbamate group or terminal urea group according to the invention and having at least two linking groups that are urethane or urea or a compound (a)(2) having at least two groups selected from carbamate groups, terminal urea groups, or combinations of the two and at least four urethane or urea linking groups. The second component (b) of the coating composition includes an acrylic polymer comprising active hydrogen-containing functional groups reactive with the third component (c). Component (c) of the coating composition is a curing agent that is reactive with the first two components. Preparation of coated articles using the compositions of the invention is also disclosed.</p>		

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CURABLE COATING COMPOSITIONS**5 CONTAINING BLENDS OF CARBAMATE-FUNCTIONAL COMPOUNDS**

The present application is a continuation-in-part of Green et al, U.S. application no. 08/997,317, filed December 23,1997, which is continuation-in-part of each
10 of the following: Green et al., U.S. application no. 08/886,321, filed July 1, 1997, which is a continuation of provisional application No. 60/021,068, filed July 1, 1996; McGee et al., U.S. application no. 08/698,529, filed August 15, 1996, which is a continuation of
15 serial no. 08/540,274, filed October 6, 1995, now abandoned; Menovcik et al., U.S. application no. 08/333,917, filed November 3, 1994, now U.S. Patent No. 5,744,550, issued April 28, 1998; U.S. application no. 08/176,608, filed Jan. 3, 1994, now abandoned; Reh fuss
20 et al., U.S. application no. 08/287,351, filed August 8, 1994, which is a continuation-in-part of serial no. 08/098,177, filed July 28, 1993, now abandoned; McGee et al., U.S. application no. 08/867,547, filed June 2, 1997, which is a continuation of U.S. application no.
25 08/513,587, filed August 10, 1995, now US Patent No. 5,726,244, issued March 10, 1998; McGee et al., U.S. application no. 08/513,587, filed August 10, 1995, now US Patent No. 5,726,244, issued March 10, 1998;
Menovcik et al., U.S. application no. 08/547,513, filed
30 October 24, 1995, now U.S. Patent No. 5,726,274, issued March 10, 1998, which is a division of U.S. application

no. 08/361,344, filed December 21, 1994, now abandoned;
Menovcik et al., U.S. application no. 08/547,174, filed
October 24, 1995, now U.S. Patent No. 5,723,552, issued
March 3, 1998, which is a divisional of U.S.

- 5 application no. 08/361,344, filed December 21, 1994,
now abandoned; Menovcik et al., U.S. application no.
08/698,524, filed August 15, 1996, now U.S. Patent No.
5,792,810, issued August 11, 1998, which is a
continuation of serial no. 08/550,880, filed October 6,
10 1995; Bammel et al., U.S. application no. 08/698,526,
filed August 15, 1996, now U.S. Patent No. 5,760,127,
issued June 2, 1998, which is a continuation of U.S.
application no. 08/686,929, filed October 6, 1995;
Ohrbom et al., U.S. application no. 08/667,261, filed
15 June 20, 1996, now U.S. Patent No. 5,777,048, issued
July 7, 1998; Ohrbom et al., U.S. application
no. 08/698,528, filed Aug. 15, 1996, now U.S. Patent No.
5,756,213, issued May 26, 1998, which is a continuation
of U.S. application no. 08/540,275, filed October 6,
20 1995, now abandoned; Ohrbom et al., U.S. application
no. 08/698,522, filed August 15, 1996, which is a
continuation of U.S. application no. 08/540,277, filed
October 6, 1995, now abandoned; McGee et al., U.S.
application no. 08/698,523, filed August 15, 1996, now
25 U.S. Patent No. 5,770,650, issued June 23, 1998, which
is a continuation of U.S. application no. 08/540,279,
filed October 6, 1995, now abandoned; Green et al.,
U.S. application no. 08/886,321, filed July 1, 1997,
which is a continuation of provisional application
30 60/021,068, filed July 1, 1996; Bammel et al., U.S.
application no. 08/831,810, filed April 2, 1997; Ohrbom

et al, U.S. application no. 08/333,804, filed Nov. 3, 1994.

Field of the Invention

This invention concerns curable coating
5 compositions, especially compositions for high-gloss topcoats, particularly for clearcoats of color-plus-clear composite coatings.

Background of the Invention

Curable, or thermosettable, coating compositions
10 are widely used in the coatings art, particularly for topcoats in the automotive and industrial coatings industry. Color-plus-clear composite coatings are particularly useful as topcoats for which exceptional gloss, depth of color, distinctness of image, or
15 special metallic effects are desired. The automotive industry has made extensive use of these coatings for automotive body panels.

Single-layer topcoats and the clearcoats of color-plus-clear composite coatings, however, require an
20 extremely high degree of clarity and gloss to achieve the desired visual effect. Such coatings also require a low degree of visual aberrations at the surface of the coating in order to achieve the desired visual effect such as high distinctness of image (DOI). As
25 such, these coatings are especially susceptible to a phenomenon known as environmental etch. Environmental etch manifests itself as spots or marks on or in the finish of the coating that often cannot be rubbed out. It is often difficult to predict the degree of
30 resistance to environmental etch that a high gloss topcoat or color-plus-clear composite coating will

exhibit. Many coating compositions known for their durability and/or weatherability when used in exterior paints, such as known high-solids enamels, do not provide the desired level of resistance to
5 environmental etch when used in high gloss coatings such as the clearcoat of a color-plus-clear composite coating.

Various compositions have been proposed to meet the above requirements for use as the topcoat coating
10 or as the clearcoat of a color-plus-clear composite coating, including polyurethanes, acid-epoxy systems and the like. However, many prior art systems suffer from disadvantages such as coatability problems, marginal compatibility with the pigmented basecoat,
15 solubility problems, and marginal appearance. Moreover, while one-pack compositions are preferred to two-pack compositions (in which the reactive component must be separated before application to prevent premature reaction), very few one-pack coating
20 compositions have been found that provide satisfactory resistance to environmental etch, especially in the demanding environment of automotive coatings.

In addition, it is desirable to provide coatings with a good combination of properties such as
25 durability, hardness, flexibility, and resistance to scratching, marring, solvents, and acids.

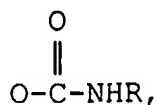
Curable coating compositions utilizing carbamate-functional resins are described, for example, in U.S. Patent Nos. 5,693,724, 5,693,723, 5,639,828, 5,512,639,
30 5,508,379, 5,451,656, 5,356,669, 5,336,566, and 5,532,061, and U.S. Application Nos. 08/886,321, filed

July 1, 1997, 08/698,529, filed August 15, 1996,
08/719,670, filed September 25, 1996, 08/166,277, filed
December 13, 1993, 08/339,999, filed November 15, 1994,
08/333,917, filed November 3, 1994, 08/176,608, filed
5 January 3, 1994, 08/287,351, filed August 8, 1994,
08/804,239, filed February 20, 1997, 08/333,804, filed
November 3, 1994, 08.884,613, filed June 30, 1997,
08/885,638, filed June 30, 1997, 08/513,587, filed
August 10, 1995, 08/867,547, filed June 2, 1997,
10 08/547,514, filed October 24, 1994, 08/547,513, filed
October 24, 1994, 08/547,174, filed October 24, 1994,
08/698,524, filed August 15, 1996, 08/698,526, filed
August 15, 1996, 08/667,261, filed June 20, 1996,
08/698,528, filed August 15, 1996, 08/698,522, filed
15 August 15, 1996, 08/698,572, filed August 15, 1996,
08/698,523, filed August 15, 1996, 08/673,935, filed
July 1, 1996, 08/886,321, filed July 1, 1997, and
08/831,810, filed April 2, 1997, each of which is
incorporated herein by reference. These coating
20 compositions can provide significant etch advantages
over other coating compositions, such as hydroxy-
functional acrylic/melamine coating compositions. It
may often be desirable, however, to provide still
further improvements in the above-described coating
25 properties.

Summary of the Invention

The present invention provides a curable coating
composition that includes at least three components: a
30 component (a), a component (b), and a component (c).

Component (a) includes one or both of two compounds, compounds (a)(1) and (a)(2) having at least one carbamate group or terminal urea group. When used in connection with the invention, the term "carbamate group" refers to a group having a structure



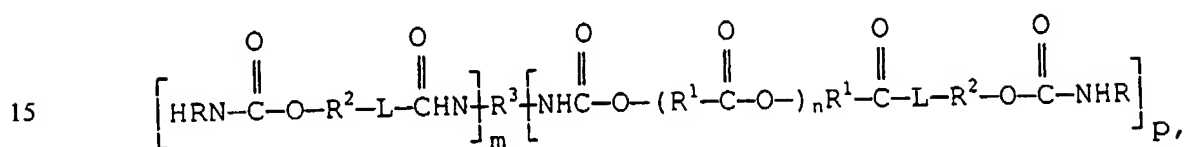
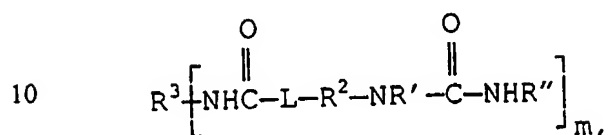
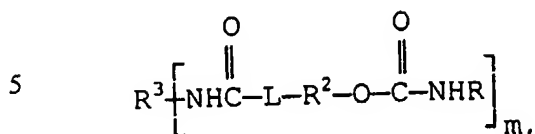
in which R is H or alkyl. Preferably, R is H or alkyl of from 1 to about 4 carbon atoms, and more preferably R is H. When used in connection with the invention, "terminal urea group" refers to a group having a structure



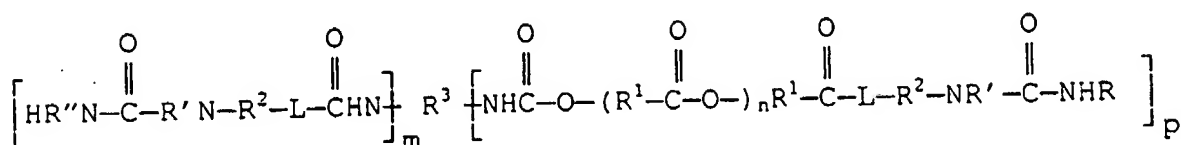
in which R' and R'' are each independently H or alkyl, or R' and R'' together form a heterocyclic ring structure. Preferably, R' and R'' are each independently H or alkyl of from 1 to about 4 carbon atoms or together form an ethylene bridge, and more preferably R' and R'' are each independently H. The terminal urea group of the invention is distinguished from urea linking groups for which R'' would be other than alkyl.

Compound (a)(1) has at least one carbamate group or terminal urea group and at least two linking groups that are urethane or urea. Preferred compounds (a)(1) may be

represented by any of the structures



and



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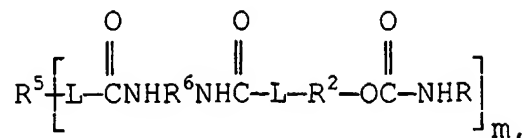
30

in which R, R', and R'' are as previously defined; R¹ is alkylene or arylalkylene, preferably alkylene, and particularly alkylene of 5 to 10 carbon atoms; R² is alkylene or substituted alkylene, preferably having from about 2 to about 4 carbon atoms; R³ is alkylene (including cycloalkylene), alkylarylene, arylene, or a structure that includes a cyanuric ring, a urethane group, a urea group, a carbodiimide group, a biuret

structure, or an allophonate group, preferably alkylene (including cycloalkylene) or a structure that includes a cyanuric ring; n is from 0 to about 10, preferably from 0 to about 5; m is from 2 to about 6, preferably 2 or 3; and L is O, NH, or NR⁴, where R⁴ is an alkyl, preferably an alkyl of 1 to about 6 carbon atoms; p is from 1 to 5, preferably 1 or 2, and m + p is 2 to 6, preferably about 3. Preferably, R³ is alkylene (including cycloalkylene), alkylarylene, arylene, or a structure that includes a cyanuric ring.

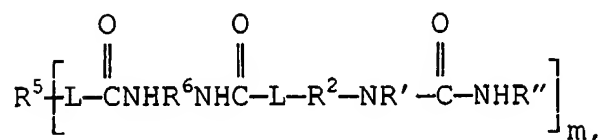
The compound (a) (1) may be prepared by a process having a step of reacting together a monomeric polyisocyanate (a) (1) (B) and a compound (a) (1) (A) having a carbamate or terminal urea group or a group that can be converted to a carbamate or terminal urea group and also having a group that is reactive with isocyanate functionality. In the case of a group that can be converted to carbamate or urea, the conversion to the carbamate or terminal urea group is carried out either at the same time as the reaction involving the polyisocyanate or afterwards to form the compound (a) (1) of the first component.

Alternatively or in addition to this compound (a) (1), the component (a) may include a compound (a) (2) having at least two groups selected from carbamate groups, terminal urea groups, or combinations of the two and at least four urethane or urea linking groups. Preferred compounds (a) (2) may be represented by any of the structures



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and



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in which R, R', and R'' are as previously defined; R¹ is
 15 alkylene or arylalkylene, preferably alkylene, and
 particularly alkylene of 5 to 10 carbon atoms; R² is
 alkylene or substituted alkylene, preferably having
 from about 2 to about 4 carbon atoms; R³ is alkylene
 (including cycloalkylene), alkylarylene, arylene, or a
 20 structure that includes a cyanuric ring, a urethane
 group, a urea group, a carbodiimide group, a biuret
 structure, or an allophonate group, preferably alkylene
 (including cycloalkylene) or a structure that includes
 a cyanuric ring; n is from 0 to about 10, preferably
 25 from 0 to about 5; m is from 2 to about 6, preferably 2
 or 3; and L is O, NH, or NR⁴, where R⁴ is an alkyl,
 preferably an alkyl of 1 to about 6 carbon atoms; p is
 from 1 to 5, preferably 1 or 2, and m + p is 2 to 6,
 preferably about 3. Preferably, R³ is alkylene
 30 (including cycloalkylene), alkylarylene, arylene, or a
 structure that includes a cyanuric ring.

The compound (a) (2) may be prepared by a synthesis
 that involves a step of reacting together a compound
 (a) (2) (A) comprising a carbamate or terminal urea group

or a group that can be converted to a carbamate or terminal urea group and also having an isocyanate group and a compound (a) (2) (B) having at least two groups reactive with isocyanate functionality. When the
5 compound (a) (2) (A) comprises a group that can be converted to a carbamate or terminal urea group, the conversion to carbamate or urea may take place at the same time as the reaction with the compound having at least two groups reactive with isocyanate functionality
10 or after that reaction is completed, to generate the compound (a) (2).

The second component (b) of the coating composition is a polymer resin that includes a polyester, a polyurethane, and/or a polyester-
15 polyurethane resin, with the resin or resins used for component (b) comprising active hydrogen-containing functional groups reactive with the third component (c).

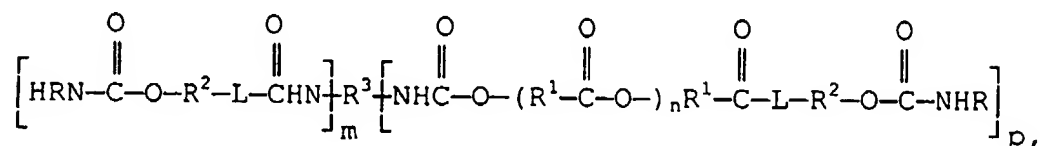
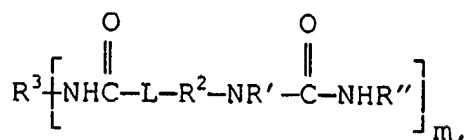
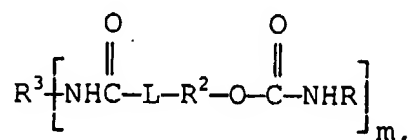
The third component (c) of the coating composition is a curing agent that is reactive with the first two
20 components.

The invention further provides an article having a substrate, in particular a flexible substrate, upon which substrate is a cured coating derived from a coating composition according to the invention and a
25 method of producing such a coating on a substrate.

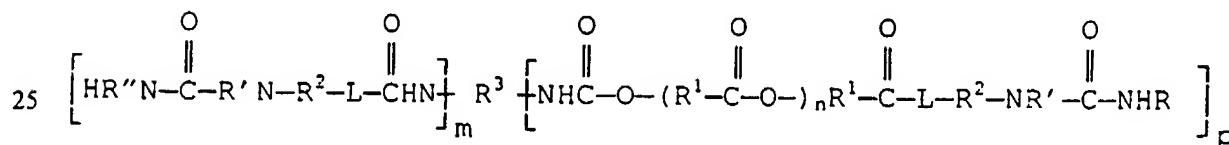
Detailed Description

The composition according to the present invention includes as a first component (a) a compound having at
30 least one carbamate group or terminal urea group. First, the component (a) may include a compound (a) (1)

having at least one carbamate group or terminal urea group and having at least two linking groups that are urethane or urea. Preferred compounds (a) (1) may be represented by any of the structures

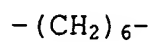


and



in which R, R', R'', R², R³, L, and m are as previously defined; p is from 1 to 5, preferably 1 or 2, and m + p is 2 to 6, preferably about 3. Preferably, R³ is alkylene (including cycloalkylene), alkylarylene, arylene, or a structure that includes a cyanuric ring.

In one preferred embodiment R^3 includes a member selected from the group of



(m + p being 2),

5

10

(m + p being 3),

15

20

(m + p being 2),

25

(m + p being 3),

30

and mixtures thereof. L is particularly preferably an oxygen atom.

The compound (a)(1) may be prepared by a step of reacting a mixture that includes at least a
5 polyisocyanate (a)(1)(B) and a compound (a)(1)(A) having a carbamate or terminal urea group as defined for the invention, or a group that can be converted to a carbamate or terminal urea group, and also having a group that is reactive with isocyanate functionality.

10 Compound (a)(1)(A) preferably has a carbamate or terminal urea group, more preferably a carbamate group. The structures for carbamate and terminal urea groups, as those terms are used in connection with the present invention, have been set out above. Alternatively,
15 compound (a)(1)(A) may have a group that can be converted to a carbamate or terminal urea group. In the case when compound (a)(1)(A) has a group that can be converted to carbamate or terminal urea, the conversion to the carbamate or terminal urea group is
20 carried out either at the same time as the reaction involving the polyisocyanate (a)(1)(B) or afterwards to form the compound (a)(1). Groups that can be converted to carbamate include cyclic carbonate groups, epoxy groups, and unsaturated bonds. Cyclic carbonate groups
25 can be converted to carbamate groups by reaction with ammonia or a primary amine, which ring-opens the cyclic carbonate to form a β -hydroxy carbamate. Epoxy groups can be converted to carbamate groups by first converting to a cyclic carbonate group by reaction with
30 CO₂. This can be done at any pressure from atmospheric

up to supercritical CO₂ pressures, but is preferably under elevated pressure (e.g., 60-150 psi). The temperature for this reaction is preferably 60-150°C. Useful catalysts include any that activate an oxirane ring, such as tertiary amine or quaternary salts (e.g., tetramethyl ammonium bromide), combinations of complex organotin halides and alkyl phosphonium halides (e.g., (CH₃)₃SnI, Bu₄SnI, Bu₄PI, and (CH₃)₄PI), potassium salts (e.g., K₂CO₃, KI) preferably in combination with crown ethers, tin octoate, calcium octoate, and the like. The cyclic carbonate group can then be converted to a carbamate group as described above. Any unsaturated bond can be converted to a carbamate group by first reacting with peroxide to convert to an epoxy group, then with CO₂ to form a cyclic carbonate, and then with ammonia or a primary amine to form the carbamate.

Other groups, such as hydroxyl groups or isocyanate groups can also be converted to carbamate groups. However, if hydroxyl groups were to be present on the compound and it is desired to convert those groups to carbamate after the reaction with the polyisocyanate (a)(1)(B), they would have to be blocked or protected so that they would not react during the reaction with compound (a)(1)(B) or else in stoichiometric excess so that some would be expected to remain unreacted for later conversion to the carbamate or terminal urea group(s). Conversion to carbamate or urea could also be carried out prior to the reaction with compound (a)(1)(A). Hydroxyl groups can be

converted to carbamate groups by reaction with a monoisocyanate (e.g., methyl isocyanate) to form a secondary carbamate group (that is, a carbamate of the structure above in which R is alkyl) or with cyanic acid (which may be formed in situ by thermal decomposition of urea) to form a primary carbamate group (i.e., R in the above formula is H). This reaction preferably occurs in the presence of a catalyst as is known in the art. A hydroxyl group can also be reacted with phosgene and then ammonia to form a primary carbamate group, or by reaction of the hydroxyl with phosgene and then a primary amine to form a compound having secondary carbamate groups. Another approach is to react an isocyanate with a compound such as hydroxyalkyl carbamate to form a carbamate-capped isocyanate derivative. For example, one isocyanate group on toluene diisocyanate can be reacted with hydroxypropyl carbamate, followed by reaction of the other isocyanate group with an excess of polyol to form a hydroxy carbamate. Finally, carbamates can be prepared by a transesterification approach where hydroxyl group is reacted with an alkyl carbamate (e.g., methyl carbamate, ethyl carbamate, butyl carbamate) to form a primary carbamate group-containing compound. This reaction is performed at elevated temperatures, preferably in the presence of a catalyst such as an organometallic catalyst (e.g., dibutyltin dilaurate). Other techniques for preparing carbamates are also known in the art and are described, for example, in P. Adams & F. Baron, "Esters of Carbamic Acid", Chemical Review, v. 65, 1965.

Groups such as oxazolidone can also be converted to terminal urea groups. For example, hydroxyethyl oxazolidone can be used to react with the polyisocyanate (a)(1)(B), followed by reaction of ammonia or a primary amine with the oxazolidone to generate the terminal urea functional group.

In addition to the carbamate or terminal urea group or the group that can be converted to a carbamate or terminal urea group, the compound (a)(1)(A) also has a group that is reactive with isocyanate functionality. Suitable groups that are reactive with isocyanate functionality include, without limitation, hydroxyl groups, primary amine groups, and secondary amine groups. Preferably, the compound (a)(1)(A) has hydroxyl groups or primary amine groups as the groups reactive with isocyanate functionality, and more preferably hydroxyl groups. The compound (a)(1)(A) has at least one group that is reactive with isocyanate functionality, and preferably it has from 1 to about 3 of such groups, and more preferably it has one such reactive group. In a preferred embodiment, the compound (a)(1)(A) has a carbamate group and a hydroxyl group. One preferred example of such a compound is a hydroxyalkyl carbamate, particularly a β -hydroxyalkyl carbamate. In another preferred embodiment, the compound (a)(1) has a terminal urea group and a hydroxyl group.

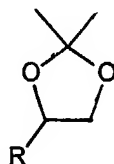
Suitable compounds (a)(1)(A) include, without limitation, any of those compounds having a carbamate or terminal urea group and a hydroxyl or primary or

secondary amine group. Illustrative examples of suitable compounds of this type include, without limitation, hydroxy alkyl carbamates and hydroxyalkylene alkyl ureas, such as hydroxyethyl carbamate, hydroxypropyl carbamate, and hydroxyethylene ethyl urea. Hydroxypropyl carbamate and hydroxyethylene urea, for example, are well known and commercially available. Amino carbamates are described in U.S. Patent 2,842,523. Compounds with hydroxyl and terminal urea groups may also be prepared by reacting the amine group of an amino alcohol with hydrochloric acid and then urea to form a hydroxy terminal urea compound. An amino alcohol can be prepared, for example, by reacting an oxazolidone with ammonia.

Amino terminal urea compounds can be prepared, for example, by reacting a ketone with a diamine having one amine group protected from reaction (e.g., by steric hindrance), followed by reaction with HNCO (e.g., as generated by thermal decomposition of urea), and finally reaction with water. Alternatively, these compounds can be prepared by starting with a compound having the group that can be converted to carbamate or terminal urea, which groups are described below, and converting that group to the carbamate or urea prior to beginning the reaction with the polyisocyanate (a)(1)(B).

Other suitable compounds (a)(1)(A) include those compounds (a)(1)(A) having a group reactive with the polyisocyanate (a)(1)(B) and a group that can be converted to carbamate such as hydroxyalkyl cyclic carbonates. Certain hydroxyalkyl cyclic carbonates

like 3-hydroxypropyl carbonate (i.e., glycerine carbonate) are commercially available. Cyclic carbonate compounds can be synthesized by any of several different approaches. One approach involves
5 reacting an epoxy group-containing compound with CO₂ under conditions and with catalysts as described hereinabove. Epoxides can also be reacted with β -butyrolactone in the presence of such catalysts. In another approach, a glycol like glycerine is reacted at
10 temperatures of at least 80°C with diethyl carbonate in the presence of a catalyst (e.g., potassium carbonate) to form a hydroxyalkyl carbonate. Alternatively, a functional compound containing a ketal of a 1,2-diol having the structure:



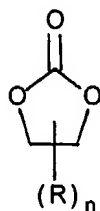
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can be ring-opened with water, preferably with a trace amount of acid, to form a 1,2-glycol, the glycol then being further reacted with diethyl carbonate to form the cyclic carbonate.

20

Cyclic carbonates typically have 5- or 6-membered rings, as is known in the art. Five-membered rings are preferred, due to their ease of synthesis and greater degree of commercial availability. Six-membered rings can be synthesized by reacting phosgene with 1,3-
25 propanediol under conditions known in the art for the formation of cyclic carbonates. Preferred hydroxyalkyl

cyclic carbonates used in the practice of the invention can be represented by the formula:



in which R (or each instance of R if n is more than 1)
 5 is a hydroxyalkyl group of 1-18 carbon atoms,
 preferably 1-6 carbon atoms, and more preferably 1-3
 carbon atoms, which may be linear or branched and may
 have substituents in addition to the hydroxyl group,
 and n is 1 or 2, which may be substituted by one or
 10 more other substituents such as blocked amines or
 unsaturated groups. The hydroxyl group may be on a
 primary, secondary, or tertiary carbon. More
 preferably, R is $-(CH_2)_p-OH$, where the hydroxyl may be
 on a primary or secondary carbon and p is 1 to 8, and
 15 even more preferably in which the hydroxyl is on a
 primary carbon and p is 1 or 2.

The second component of the reaction mixture used
 in forming compound (a)(1) is a polyisocyanate
 (a)(1)(B). Suitable examples of polyisocyanate
 20 compounds include both aliphatic polyisocyanates and
 aromatic polyisocyanates. Useful polyisocyanates
 include monomeric isocyanates, for example aliphatic
 diisocyanates such as ethylene diisocyanate, 1,2-
 diisocyanatopropane, 1,3-diisocyanatopropane, 1,6-
 25 diisocyanatohexane (hexamethylene diisocyanate or
 HMDI), 1,4-butylene diisocyanate, lysine diisocyanate,

1,4-methylene bis-(cyclohexyl isocyanate) and isophorone diisocyanate (IPDI), and aromatic diisocyanates and arylaliphatic diisocyanates such as the various isomers of toluene diisocyanate, meta-xylylenediisocyanate and para-xylylenediisocyanate, 4-chloro-1,3-phenylene diisocyanate, 1,5-tetrahydronaphthalene diisocyanate, 4,4'-dibenzyl diisocyanate, and 1,2,4-benzene triisocyanate. In addition, the various isomers of $\alpha,\alpha,\alpha',\alpha'$ -tetramethyl xylylene diisocyanate can be used. Isocyanate-functional oligomers or low molecular weight reaction products of the monomeric isocyanates, which may have from 2 to about 6 isocyanate groups, may also be used. Examples of these include isocyanurates and the reaction products of excess isocyanate with polyols, such as the product of three moles of diisocyanate with a mole of a triol (e.g., 3 moles of IPDI with one mole of trimethylolpropane or two moles of IPDI with one mole of neopentyl glycol); reaction products of isocyanate with urea (biurets); and reaction products of isocyanate with urethane (allophanates). The polyisocyanate preferably has two to four isocyanate groups, and more preferably the polyisocyanate has 2 or 3 isocyanate groups per molecule. Isocyanurates such as the isocyanurates of isophorone diisocyanate or hexamethylene diisocyanate are particularly preferred.

In one preferred embodiment, the polyisocyanate (a)(1)(B) is isophorone diisocyanate, the isocyanurate of isophorone diisocyanate, hexamethylene diisocyanate, the isocyanurate of isophorone diisocyanate, or a

combination of these. In another preferred embodiment, the polyisocyanate is an isocyanate-functional monomeric or oligomeric, preferably monomeric, reaction product of a diisocyanate and a polyol. Such a
5 reaction product may prepared by reacting one mole of a diisocyanate per equivalent of polyol. This endcapping is preferably accomplished by reacting at least two equivalents of isocyanate of a diisocyanate for each equivalent of hydroxyl of the polyol. The diisocyanate
10 is preferably isophorone diisocyanate or hexamethylene diisocyanate. The polyol is preferably 2-ethyl-1,6-hexanediol, trimethylolpropane, neopentyl glycol, or a combination of these.

In one preferred embodiment, the compound (a) (1)
15 is produced by a step that includes reacting a mixture of an isocyanate (preferably a diisocyanate, e.g., HDI, IPDI, or the isocyanate-functional endcapped polyol described in the previous paragraph) and a compound such as hydroxypropyl carbamate to form a carbamate-
20 capped polyisocyanate derivative, as described in U.S. Patent No. 5,512,639.

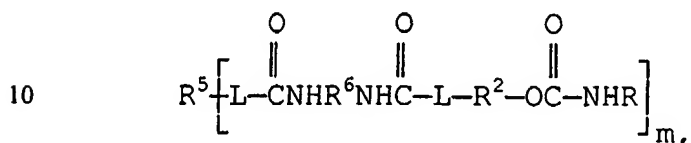
Compound (a) (1) may also be formed by a reaction mixture includes, in addition to compounds (a) (1) (A) and (a) (1) (B), a compound (a) (1) (C) that is an active-
25 hydrogen chain extension agent. Chain extension agents may be used to increase the length of compound (a) (1) or to bridge together two or more products of the reaction of compounds (a) (1) (A) and (a) (1) (B). Useful active hydrogen-containing chain extension agents
30 generally contain at least two, preferably about two, active hydrogen groups, for example, diols, dithiols,

diamines, or compounds having a mixture of hydroxyl, thiol, and amine groups, such as alkanolamines, aminoalkyl mercaptans, and hydroxyalkyl mercaptans, among others. For purposes of this aspect of the invention, both primary and secondary amine groups are considered as having one active hydrogen. Active hydrogen-containing chain extension agents also include water. In a preferred embodiment of the invention, a polyol is used as the chain extension agent. In an especially preferred embodiment, a diol is used as the chain extension agent with little or no higher polyols, so as to minimize branching. Examples of preferred compounds (a)(1)(C) include, without limitation, 1,6-hexanediol, 1,2-hexanediol, 2-ethyl-1,3-hexanediol, 2-ethyl-1,6-hexanediol, 3-hydroxy-2,2-dimethylpropyl 3-hydroxy-2,2-dimethylpropionate (sold by Eastman Chemical Co. as Esterdiol 204), 1,4-butanediol, 1,5-pentanediol, neopentyl glycol, cyclohexanedimethanol (sold as CHDM by Eastman Chemical Co.), ethyl-propyl-1,5-pentanediol, 2-methyl-2,4-pentanediol, 2,2,4-trimethyl-1,3-pentanediol, 2,4,7,9-tetramethyl-5-decyn-4,7-diol, 1,3-dihydroxyacetone dimer, 2-butene-1,4-diol, pantothenol, dimethyltartrate, pentaethylene glycol, dimethyl silyl dipropanol, and 2,2'-thiodiethanol. While polyhydroxy compounds containing at least three hydroxyl groups may be used as chain extenders, the use of these compounds may produce higher molecular weight, more branched compounds. Higher-functional polyhydroxy compounds include, for example, trimethylolpropane, trimethylolethane, pentaerythritol, among other compounds. In a

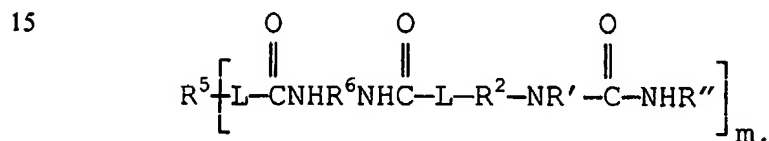
particularly preferred embodiment, the monomeric isocyanate is a diisocyanate, especially isophorone or hexamethylene diisocyanate and an average of one of the isocyanate groups per molecule is reacted with the compound (a)(1)(A) comprising a group that is reactive with isocyanate and a carbamate group or group that can be converted into a carbamate group, preferably with hydroxypropyl carbamate, and the remaining isocyanate groups are reacted with a polyol, particularly with 2-ethyl-1,6-hexanediol. The reactions of the polyisocyanate with compounds (a)(1)(A) and (a)(1)(C) can be carried out in any order, including concurrently. While a mixture of reaction products may be expected each of the isocyanate groups has about the same reactivity, at least a part should be the idealized product in which a molecule of the polyisocyanate has reacted with both a compound (a)(1)(A) and a compound (a)(1)(C).

Another method of synthesis of compound (a)(1) is to first react the isocyanate groups of a polyisocyanate with a compound having a group that is reactive with isocyanate and also a non-isocyanate functional group. This adduct is then reacted with a compound comprising at least one carbamate group or group that can be converted to carbamate and at least one group reactive with the non-NCO functional groups. Examples of non-isocyanate functional groups include carboxyl, epoxy, hydroxyl, amino. Suitable examples of methods for converting such groups to carbamate or urea groups have already been described above in detail.

Instead of or in addition to the compound (a) (1), component (a) may include a compound (a) (2) having at least two carbamate and/or terminal urea groups and at least four urethane or urea linking groups. Preferred
 5 compounds (a) (2) may be represented by any of the structures



and



20 in which R, R', R'', R², L, and m are as previously defined; R⁵ and R⁶ are each independently alkylene (including cycloalkylene), preferably having from 1 to about 18 carbon atoms, particularly preferably from about 5 to about 12 carbon atoms, alkylarylene, or
 25 arylene, or R⁶ is a structure that includes a cyanuric ring, a biuret structure, or an allophonate group.

Compound (a) (2) may be prepared by a step of reacting together a compound (a) (2) (A) comprising a carbamate or terminal urea group or a group that can be
 30 converted to a carbamate or terminal urea group and also having an isocyanate group and a second compound (a) (2) (B) having at least two groups reactive with isocyanate functionality.

The compound (a) (2) (A) preferably has, on average, one isocyanate group per molecule. The compound (a) (2) (A) also preferably has a carbamate or terminal urea group, and particularly preferably has a carbamate group. In one embodiment, compound (a) (2) (A) is a reaction product of an hydroxyalkyl carbamate and a polyisocyanate compound. In another embodiment, compound (a) (2) (A) is a reaction product of an hydroxyalkyl cyclic carbonate and a polyisocyanate compound. Preferably, a hydroxyalkyl carbamate and a polyisocyanate are reacted to produce an isocyanate-functional compound with carbamate functionality. In a particularly preferred embodiment, a β -hydroxyalkyl carbamate is reacted with one of the isocyanate groups of a diisocyanate, such as IPDI or HMDI, or with one or two of the isocyanate groups of an isocyanurate, such as the isocyanurate of HMDI or IPDI, to produce a compound (a) (2) (A) having both isocyanate and carbamate functionality.

The compound (a) (2) (B) having at least two groups reactive with isocyanate functionality may be a diamine or a polyol, preferably a diol. Particularly preferred compounds (a) (2) (B) having at least two groups reactive with isocyanate functionality include linear and branched diols such as 1,6-hexanediol, 2-ethyl-1,6-hexanediol, and neopentyl glycol.

Component (b) of the coating compositions of the invention includes an acrylic polymer. The acrylic polymer comprises active hydrogen-containing functional groups that are reactive with the third component (c)

curing agent. Suitable active hydrogen-containing functional groups include, without limitation, hydroxyl functionality, acid functionality, carbamate functionality, terminal urea functionality, and combinations of these. In a preferred embodiment, the acrylic polymer (b) has carbamate or terminal urea functionality. The carbamate or terminal urea functionality may be introduced to the polymer by either polymerizing using a carbamate- or terminal urea-functional monomer or by reacting a functional group on the formed polymer in a further reaction to produce a carbamate or terminal urea functionality at that position. If the functional group on the acrylic polymer (b) is an isocyanate group, the group can be reacted with a hydroxyalkyl carbamate, or with a hydroxy-containing epoxide with the epoxy group subsequently converted to carbamate by reaction with CO₂ and then ammonia. Preferably, an isocyanate-functional acrylic polymer is reacted with hydroxyethyl carbamate, hydroxypropyl carbamate, hydroxybutyl carbamate, or mixtures thereof. If the functional group is hydroxyl, the reactive group on the carbamate-containing compound may be oxygen of the C(=O)O portion of the carbamate group on an alkyl carbamate or methylol, such as with methylol acrylamide (HO-CH₂-NH-C(=O)-CH=CH₂). In the case of the C(=O)O group on an alkyl carbamate, the hydroxyl group on the polymer undergoes a transesterification with the C(=O)O group, resulting in the carbamate group being appended to the polymer. In the case of methylol acrylamide,

the unsaturated double bond is then reacted with peroxide, CO₂, and ammonia as described above. If the functional group on the polymer is a carboxyl group, the acid group can be reacted with epichlorohydrin to
5 form a monoglycidyl ester, which can be converted to carbamate by reaction with CO₂, and then ammonia.

Carbamate functionality can also be introduced to the acrylic polymer of component (b) by reacting the polymer with a compound that has a group that can be converted to a
10 carbamate, and then converting that group to the carbamate. Examples of suitable compounds with groups that can be converted to a carbamate include, without limitation, active hydrogen-containing cyclic carbonate compounds (e.g., the reaction product of glycidol and CO₂) that are convertible
15 to carbamate by reaction with ammonia, monoglycidyl ethers and esters convertible to carbamate by reaction with CO₂ and then ammonia, allyl alcohols where the alcohol group is reactive with isocyanate functionality and the double bond can be converted to carbamate by reaction with peroxide, and
20 vinyl esters where the ester group is reactive with isocyanate functionality and the vinyl group can be converted to carbamate by reaction with peroxide, then CO₂, and then ammonia. Any of the above compounds can be utilized as compounds containing carbamate groups rather
25 than groups convertible to carbamate by converting the group to carbamate prior to reaction with the polymer.

The acrylic resin preferably has a molecular weight of 500 to 1,000,000, and more preferably of 1500 to 50,000. As used herein, "molecular weight" refers
30 to number average molecular weight, which may be

determined by the GPC method using a polystyrene standard. Such polymers can be prepared from acrylic monomers such as methyl acrylate, acrylic acid, methacrylic acid, methyl methacrylate, butyl
5 methacrylate, cyclohexyl methacrylate, and the like. The functional group can be incorporated into the ester portion of the acrylic monomer. For example, hydroxy-functional acrylic monomers that can be used to form such polymers include hydroxyethyl acrylate,
10 hydroxybutyl acrylate, hydroxybutyl methacrylate, hydroxypropyl acrylate, and the like; amino-functional acrylic monomers would include t-butylaminoethyl methacrylate and t-butylamino-ethylacrylate; acid-functional monomers would include acrylic acid,
15 methacrylic acid, and itaconic acid; epoxide-functional monomers would include glycidyl acrylate and glycidyl methacrylate; and so on.

One way to prepare such polymers is to prepare an acrylic monomer having a carbamate functionality in the
20 ester portion of the monomer. Such monomers are well-known in the art and are described, for example in U.S. Patents 3,479,328, 3,674,838, 4,126,747, 4,279,833, and 4,340,497, 5,356,669, and WO 94/10211, the disclosures of which are incorporated herein by reference. One
25 method of synthesis involves reaction of a hydroxy-functional monomer with cyanic acid (which may be formed by the thermal decomposition of urea) to form the carbamyloxy carboxylate (i.e., carbamate-modified (meth)acrylate). Another method of synthesis reacts an
30 α,β -unsaturated acid ester with a hydroxy carbamate

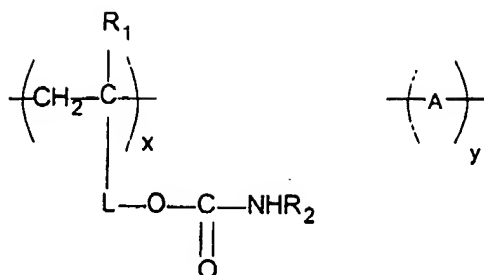
ester to form the carbamyloxy carboxylate. Yet another technique involves formation of a hydroxyalkyl carbamate by reacting a primary or secondary amine or diamine with a cyclic carbonate such as ethylene carbonate. The hydroxyl group on the hydroxyalkyl carbamate is then esterified by reaction with acrylic or methacrylic acid to form the monomer. Other methods of preparing carbamate-modified acrylic monomers are described in the art, and can be utilized as well. The acrylic monomer can then be polymerized along with other ethylenically-unsaturated monomers, if desired, by techniques well-known in the art.

An alternative route for preparing a carbamate-functional polymer is to react an already-formed polymer such as an acrylic polymer with another component to form a carbamate-functional group appended to the polymer backbone, as described in U.S. Patent 4,758,632, the disclosure of which is incorporated herein by reference. One technique for preparing acrylic polymers useful as the second component involves thermally decomposing urea (to give off ammonia and HNCO) in the presence of a hydroxy-functional acrylic polymer to form a carbamate-functional acrylic polymer. Another technique involves reacting the hydroxyl group of a hydroxyalkyl carbamate with the isocyanate group of an isocyanate-functional acrylic or vinyl monomer to form the carbamate-functional acrylic. Isocyanate-functional acrylics are known in the art and are described, for example in U.S. Patent 4,301,257, the disclosure of which is incorporated herein by reference. Isocyanate vinyl

monomers are well-known in the art and include unsaturated m-tetramethyl xylene isocyanate and isocyanatoethyl methacrylate. Yet another technique is to react the cyclic carbonate group on a cyclic carbonate-functional acrylic with ammonia in order to form the carbamate-functional acrylic. Cyclic carbonate-functional acrylic polymers are known in the art and are described, for example, in U.S. Patent 2,979,514, the disclosure of which is incorporated herein by reference. Another technique is to transcarbamylate a hydroxy-functional acrylic polymer with an alkyl carbamate. A more difficult, but feasible way of preparing the polymer would be to trans-esterify an acrylate polymer with a hydroxyalkyl carbamate.

Modified acrylics can also be used as the polymer (b) according to the invention. Such acrylics may be polyester-modified acrylics or polyurethane-modified acrylics, as is well-known in the art. Polyester-modified acrylics modified with ϵ -caprolactone are described in U.S. Patent 4,546,046 of Etzell et al., the disclosure of which is incorporated herein by reference. Polyurethane-modified acrylics are also well-known in the art. They are described, for example, in U.S. Patent 4,584,354, the disclosure of which is incorporated herein by reference.

A carbamate-functional addition polymer component can be represented by the randomly repeating units according to the following formula:



In the above formula, R₁ represents H or CH₃. R₂ represents H, alkyl, preferably of 1 to 6 carbon atoms, or cycloalkyl, preferably up to 6 ring carbon atoms.

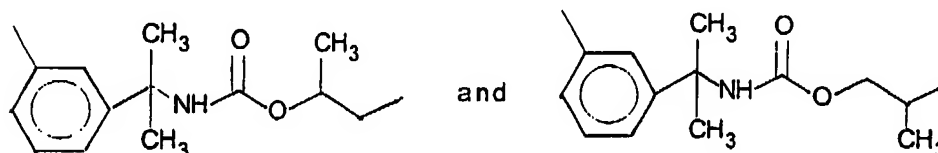
5 It is to be understood that the terms alkyl and cycloalkyl are to include substituted alkyl and cycloalkyl, such as halogen-substituted alkyl or cycloalkyl. Substituents that will have an adverse impact on the properties of the cured material, however, are to be avoided. For example, ether linkages are thought to be susceptible to hydrolysis, and should be avoided in locations that would place the ether linkage in the crosslink matrix. The values x and y represent weight percentages, with x being 10 to 15 90 % and preferably 40 to 60 %, and y being 90 to 10 % and preferably 60 to 40 %.

In the formula, A represents polymerized units derived from one or more ethylenically unsaturated comonomers. Such monomers for copolymerization are 20 known in the art. They include alkyl esters of acrylic or methacrylic acid, e.g., ethyl acrylate, butyl acrylate, 2-ethylhexyl acrylate, butyl methacrylate, isodecyl methacrylate, hydroxyethyl methacrylate, hydroxypropyl acrylate, and the like; and vinyl 25 monomers such as unsaturated m-tetramethyl xylene

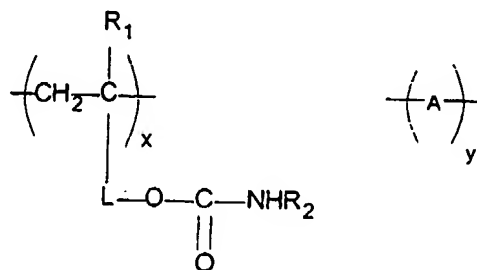
isocyanate, styrene, vinyl toluene and the like. -
 Suitable comonomers also include monomer having other
 functionalities, including hydroxyl, acid, and epoxide
 functionalities.

- 5 L represents a divalent linking group, preferably
 an aliphatic of 1 to 8 carbon atoms, cycloaliphatic, or
 aromatic linking group of 6 to 10 carbon atoms.

Examples of L include



- 10 $-(CH_2)-$, $-(CH_2)_2-$, $-(CH_2)_4-$, and the like. In one
 preferred embodiment, $-L-$ is represented by $-COO-L'-$
 where L' is a divalent linking group. Thus, in a
 preferred embodiment of the invention, the polymer
 component (a) is represented by randomly repeating
 15 units according to the following formula:



- In this formula, R_1 , R_2 , A , x , and y are as defined
 above. L' may be a divalent aliphatic linking group,
 preferably of 1 to 8 carbon atoms, e.g., $-(CH_2)-$,
 20 $-(CH_2)_2-$, $-(CH_2)_4-$, and the like, or a divalent
 cycloaliphatic linking group, preferably up to 8 carbon
 atoms, e.g., cyclohexyl, and the like. However, other

divalent linking groups can be used, depending on the technique used to prepare the polymer. For example, if a hydroxyalkyl carbamate is adducted onto an isocyanate-functional acrylic polymer, the linking
5 group L' would include an -NHC(=O)O- urethane linkage as a residue of the isocyanate group.

As a third component (c), the coating composition includes a curing agent or crosslinker that is reactive with the first two components, (a) and (b). The curing
10 agent has, on average, at least about two functional groups reactive with the first and second components. The functional groups may be of more than one kind, each kind being reactive with one or both of the first two components.

15 Useful curing agents include materials having active methylol or methylalkoxy groups, such as aminoplast crosslinking agents or phenol/formaldehyde adducts; curing agents that have isocyanate groups, particularly blocked isocyanate curing agents, curing
20 agents that have epoxide groups, amine groups, acid groups, siloxane groups, cyclic carbonate groups, and anhydride groups; and mixtures thereof. Examples of preferred curing agent compounds include, without limitation, melamine formaldehyde resin (including
25 monomeric or polymeric melamine resin and partially or fully alkylated melamine resin), blocked or unblocked polyisocyanates (e.g., TDI, MDI, isophorone diisocyanate, hexamethylene diisocyanate, and isocyanurates of these, which may be blocked for
30 example with alcohols or oximes), urea resins (e.g., methylol ureas such as urea formaldehyde resin, alkoxy

ureas such as butylated urea formaldehyde resin), polyanhydrides (e.g., polysuccinic anhydride), and polysiloxanes (e.g., trimethoxy siloxane). Another suitable crosslinking agent is tris(alkoxy carbonylamino) triazine (available from Cytec Industries under the tradename TACT). The curing agent may be combinations of these, particularly combinations that include aminoplast crosslinking agents. Aminoplast resins such as melamine formaldehyde resins or urea formaldehyde resins are especially preferred. Combinations of tris(alkoxy carbonylamino) triazine with a melamine formaldehyde resin and/or a blocked isocyanate curing agent are likewise suitable and desirable. Component (b) may also contain groups that are reactive with the carbamate group of component (a), such as an acrylic polymer containing polymerized isobutoxymethyl acrylamide groups.

A solvent may optionally be utilized in the coating composition used in the practice of the present invention. Although the composition used according to the present invention may be utilized, for example, in the form of substantially solid powder, or a dispersion, it is often desirable that the composition is in a substantially liquid state, which can be accomplished with the use of a solvent. This solvent should act as a solvent with respect to the components of the composition. In general, the solvent can be any organic solvent and/or water. In one preferred embodiment, the solvent is a polar organic solvent. More preferably, the solvent is selected from polar aliphatic solvents or polar aromatic solvents. Still

more preferably, the solvent is a ketone, ester, acetate, aprotic amide, aprotic sulfoxide, aprotic amine, or a combination of any of these. Examples of useful solvents include, without limitation, methyl
5 ethyl ketone, methyl isobutyl ketone, m-amyl acetate, ethylene glycol butyl ether-acetate, propylene glycol monomethyl ether acetate, xylene, N-methylpyrrolidone, blends of aromatic hydrocarbons, and mixtures of these. In another preferred embodiment, the solvent is water
10 or a mixture of water with small amounts of co-solvents.

The coating composition used in the practice of the invention may include a catalyst to enhance the cure reaction. For example, when aminoplast compounds,
15 especially monomeric melamines, are used as a curing agent, a strong acid catalyst may be utilized to enhance the cure reaction. Such catalysts are well-known in the art and include, without limitation, p-toluenesulfonic acid, dinonylnaphthalene disulfonic
20 acid, dodecylbenzenesulfonic acid, phenyl acid phosphate, monobutyl maleate, butyl phosphate, and hydroxy phosphate ester. Strong acid catalysts are often blocked, e.g. with an amine. Other catalysts that may be useful in the composition of the invention
25 include Lewis acids, zinc salts, and tin salts.

In a preferred embodiment of the invention, the solvent is present in the coating composition in an amount of from about 0.01 weight percent to about 99 weight percent, preferably from about 10 weight percent
30 to about 60 weight percent, and more preferably from about 30 weight percent to about 50 weight percent.

Coating compositions can be coated on the article by any of a number of techniques well-known in the art. These include, for example, spray coating, dip coating, roll coating, curtain coating, and the like. For automotive body panels, spray coating is preferred.

Additional agents, for example surfactants, fillers, stabilizers, wetting agents, dispersing agents, adhesion promoters, UV absorbers, hindered amine light stabilizers, etc. may be incorporated into the coating composition. While such additives are well-known in the prior art, the amount used must be controlled to avoid adversely affecting the coating characteristics.

The coating composition according to the invention is preferably utilized in a high-gloss coating and/or as the clearcoat of a composite color-plus-clear coating. High-gloss coatings as used herein are coatings having a 20° gloss (ASTM D523-89) or a DOI (ASTM E430-91) of at least 80.

When the coating composition of the invention is used as a high-gloss pigmented paint coating, the pigment may be any organic or inorganic compounds or colored materials, fillers, metallic or other inorganic flake materials such as mica or aluminum flake, and other materials of kind that the art normally includes in such coatings. Pigments and other insoluble particulate compounds such as fillers are usually used in the composition in an amount of 1% to 100%, based on the total solid weight of binder components (i.e., a pigment-to-binder ratio of 0.1 to 1).

When the coating composition according to the invention is used as the clearcoat of a composite color-plus-clear coating, the pigmented basecoat composition may any of a number of types well-known in the art, and does not require explanation in detail herein. Polymers known in the art to be useful in basecoat compositions include acrylics, vinyls, polyurethanes, polycarbonates, polyesters, alkyds, and polysiloxanes. Preferred polymers include acrylics and polyurethanes. In one preferred embodiment of the invention, the basecoat composition also utilizes a carbamate-functional acrylic polymer. Basecoat polymers may be thermoplastic, but are preferably crosslinkable and comprise one or more type of crosslinkable functional groups. Such groups include, for example, hydroxy, isocyanate, amine, epoxy, acrylate, vinyl, silane, and acetoacetate groups. These groups may be masked or blocked in such a way so that they are unblocked and available for the crosslinking reaction under the desired curing conditions, generally elevated temperatures. Useful crosslinkable functional groups include hydroxy, epoxy, acid, anhydride, silane, and acetoacetate groups. Preferred crosslinkable functional groups include hydroxy functional groups and amino functional groups.

Basecoat polymers may be self-crosslinkable, or may require a separate crosslinking agent that is reactive with the functional groups of the polymer. When the polymer comprises hydroxy functional groups, for example, the crosslinking agent may be an aminoplast resin, isocyanate and blocked isocyanates

(including isocyanurates), and acid or anhydride functional crosslinking agents.

The coating compositions described herein are preferably subjected to conditions so as to cure the coating layers. Although various methods of curing may be used, heat-curing is preferred. Generally, heat curing is effected by exposing the coated article to elevated temperatures provided primarily by radiative heat sources. Curing temperatures will vary depending on the particular blocking groups used in the cross-linking agents, however they generally range between 90°C and 180°C. The first compounds according to the present invention are preferably reactive even at relatively low cure temperatures. Thus, in a preferred embodiment, the cure temperature is preferably between 115°C and 150°C, and more preferably at temperatures between 115°C and 140°C for a blocked acid catalyzed system. For an unblocked acid catalyzed system, the cure temperature is preferably between 80°C and 100°C. The curing time will vary depending on the particular components used, and physical parameters such as the thickness of the layers, however, typical curing times range from 15 to 60 minutes, and preferably 15-25 minutes for blocked acid catalyzed systems and 10-20 minutes for unblocked acid catalyzed systems.

The invention is further described in the following examples. The examples are merely illustrative and do not in any way limit the scope of the invention as described and claimed. All parts are parts by weight unless otherwise noted.

Preparation 1

A mixture of 278.1 parts by weight of methyl isoamyl ketone, 1272.8 parts by weight of T1890-A (isocyanurate of isophorone diisocyanate, available from CreaNova), and 0.7 parts by weight of dibutyl tin dilaurate was heated in a suitable reaction vessel to 40°C under an inert atmosphere. Next, 463.5 parts by weight of beta-hydroxy butyl carbamate was added, followed by an additional 115.9 parts by weight of methyl isoamyl ketone. The reaction was heated to 80° and held at that temperature of about two hours. Finally, 148.3 parts by weight of isobutyl alcohol was added.

15 Preparation 2

A mixture of 333 parts by weight of isophorone diisocyanate, 300 parts by weight of methyl isoamyl ketone, and 0.5 parts by weight of dibutyl tin dilaurate was charged to a suitable reaction vessel under an inert atmosphere. An addition of 109.7 parts by weight of 2-ethyl-1,3-hexanediol was made to the flask. During the addition, the temperature was kept below 40°C. Next, 20 parts by weight of methyl isoamyl ketone was added. The temperature of the reaction mixture was increased to 50°C followed by addition of 192.9 parts by weight of beta-hydroxy butyl carbamate and an addition 421 parts by weight of methyl isoamyl ketone. The reaction mixture was slowly increased to 90°C and 40 parts by weight of isobutanol were added. The reaction mixture was held at 90°C until all of the -isocyanate functionality had reacted.

Preparation 3

An acrylic resin having cyclic carbonate functionality (from copolymerization of the cyclic carbonate of glycidyl methacrylate) and a theoretical T_g of -39°C (as determined from the Fox equation) was reacted with an excess of ammonium hydroxide. Following completion of the reaction, as determined by infrared analysis, the reaction mixture was vacuum stripped to a measured 96.7% nonvolatile content. The resulting product of a beta-hydroxy carbamate-functional acrylic had a theoretical carbamate equivalent weight of 723 grams per equivalent.

Preparation 4

An acrylic resin was prepared with a theoretical T_g of -24 , hydroxyl equivalent weight of 365 grams/equivalent, and a weight average molecular weight of 3858.

Preparation 5

A mixture of 532.9 parts by weight methyl isoamyl ketone, 597.2 parts by weight of hexamethylene diisocyanate, and 0.8 parts by weight of dibutyl tin dilaurate were heated to 31°C under an inert atmosphere. Keeping the temperature below 71°C , 259.8 parts by weight of 2-ethyl-1,3-hexanediol were then added over about 30 minutes. Following the addition, the reaction mixture was heated to 80°C and held at that temperature for about 70 minutes. An addition of 862.3 parts of T-1890/100 (an isocyanurate available

from CreaNova) was made to the reaction mixture. After the T-1890/100 was in solution, 944.3 parts by weight of beta-hydroxy butyl carbamate was added over about 17 minutes, holding the temperature of the reaction mixture below 92°C. An additional 106.6 parts of methyl isoamyl ketone was added and the reaction mixture held at 90°C until all of the isocyanate functionality was consumed. Then 264.6 parts by weight of isobutanol were added.

10

Example 1

A clearcoat coating composition was prepared by mixing together 163.2 parts by weight of the product of Preparation 5, 29.4 parts by weight of the resin of Preparation 3, 40.9 parts by weight of Resimene® 747 (a hexamethoxymethyl melamine available from Solutia Inc., Springfield MA), 6.3 parts by weight of a UVA solution, 3.0 parts by weight of a hindered amine light stabilizer, 1.6 parts by weight of a rheology control agent, 12.0 parts by weight of a blocked sulfonic acid catalyst (25% active by weight), 10.0 parts by weight of n-butanol, 14.0 parts by weight of Exxate 600 (an oxo-hexyl acetate available from Exxon Chemical Co.), and 120.8 parts by weight of methyl isoamyl ketone.

The measured nonvolatile content for the composition was 51%NV by weight. The composition produced a continuous film when sprayed over a black basecoat on a test panel.

Example 2

A clearcoat coating composition was prepared by mixing together 195.2 parts by weight of the product of Preparation 1, 29.6 parts by weight of the resin of Preparation 3, 39.7 parts by weight of Resimene® 747 (a hexamethoxymethyl melamine available from Solutia Inc., Springfield MA), 6.3 grams of a UVA solution (95% nonvolatile by weight), 3.0 parts by weight of a hindered amine light stabilizer, 1.6 parts by weight of a rheology control agent (12.5% nonvolatile by weight), 12.0 parts by weight of a blocked sulfonic acid catalyst (25% active by weight), 10.0 parts by weight of n-butanol, 14.0 parts by weight of Exxate 600 (an oxo-hexyl acetate available from Exxon Chemical Co.), and 89.8 parts by weight of methyl isoamyl ketone. The measured nonvolatile content for the composition was 51.7%NV by weight.

Comparative Example A

A clearcoat coating composition was prepared by mixing together 101.9 parts by weight of the product of Preparation 5, 20.5 parts by weight of Resimene® 747 (a hexamethoxymethyl melamine available from Solutia Inc., Springfield MA), 3.2 parts by weight of a UVA solution (95% nonvolatile by weight), 1.5 parts by weight of a hindered amine light stabilizer, 0.8 parts by weight of a rheology control agent (12.5% nonvolatile by weight), 6.0 parts by weight of a blocked sulfonic acid catalyst (25% active by weight), 0.02 parts by weight of n-butanol, 0.55 parts by weight of Exxate 600 (an oxo-hexyl acetate available from Exxon Chemical Co.), and

14.8 parts by weight of methyl isoamyl ketone. The composition did not produce a continuous film.

Comparative Example B

5 A clearcoat coating composition was prepared by mixing together 99.9 parts by weight of the product of Preparation 2, 14.6 parts by weight of Resimene® 747 (a hexamethoxymethyl melamine available from Solutia Inc., Springfield MA), 2.6 parts by weight of a UVA solution
10 (95% nonvolatile by weight), 1.2 parts by weight of a hindered amine light stabilizer, 0.6 parts by weight of a rheology control agent (12.5% nonvolatile by weight), 4.9 parts by weight of a blocked sulfonic acid catalyst (25% active by weight), 0.02 parts by weight of n-
15 butanol, and 0.45 parts by weight of Exxate 600 (an oxo-hexyl acetate available from Exxon Chemical Co.). The composition did not produce a continuous film.

Comparative Example C

20 A clearcoat coating composition was prepared by mixing together 123.1 parts by weight of the product of Preparation 1, 19.8 parts by weight of Resimene® 747 (a hexamethoxymethyl melamine available from Solutia Inc., Springfield MA), 3.2 parts by weight of a UVA solution
25 (95% nonvolatile by weight), 1.5 parts by weight of a hindered amine light stabilizer, 0.8 parts by weight of a rheology control agent (12.5% nonvolatile by weight), 6.0 parts by weight of a blocked sulfonic acid catalyst (25% active by weight), 0.02 parts by weight of n-
30 butanol, 0.55 parts by weight of Exxate 600 (an oxo-hexyl acetate available from Exxon Chemical Co.), and

14.8 parts by weight of methyl isoamyl ketone. The composition did not produce a continuous film.

Examples 3-4 and Comparative Example D

- 5 Clearcoat coating compositions were prepared by mixing together ingredients as shown in the following table.

Ingredient	Example 3	Example 4	Compar. Ex. D
Preparation 4 resin	141.3	185.6	206.3
Preparation 1 resin	151.2	50.4	--
Resimene® 747	30.9	30.9	30.9
Catalyst	8.0	8.0	8.0
Adhesion additive	4.8	4.8	4.8
Rheology control agents	32.3	32.3	32.3
Light stabilizer additives	31.7	31.7	31.7
Flow additive	2.0	2.0	2.0
Exxate 600	3.24	19.2	33.2
Methyl isoamyl ketone	--	16.0	30.0

Testing of Coating Compositions of Examples 3 and 4 and

10 Comparative Example D

- Primed plastic panels were coated with a black basecoat and then the clearcoat composition wet-on-wet. The coated panels were cured after application of the clearcoat composition by baking at 120°C for 30 minutes
- 15 in a gas-fired oven. The cured clear coat films were 1.5-2.0 mils thick.

The coated panels were subjected to outdoor exposure testing in a Jacksonville, FL environmental

etch testing program. The amount of environmental etch was rated on a scale of 1 to 10D, with 1 being no or little film damage up to 10 being severe film damage and 10A, 10B, 10C, and 10D being increasingly severe
5 film damage.

Clearcoat Composition	Etch Rating
Example 3	6
Example 4	4
Comparative Ex. D	9

The invention has been described in detail with reference to preferred embodiments thereof. It should
10 be understood, however, that variations and modifications can be made within the spirit and scope of the invention.

What is claimed is:

1. A curable coating composition, comprising
(a) a component selected from the group consisting of:

- 5 (1) a compound having at least one carbamate group or urea group that is prepared by a step of reacting a mixture comprising:
(A) a compound comprising a carbamate or urea group or a group that can be converted to a carbamate or urea group
10 and a group that is reactive with (a) (1) (B) and
(B) a polyisocyanate,
(2) a compound having at least one carbamate group or urea group that is prepared by a
15 step of reacting a mixture comprising:
(A) a compound comprising a carbamate or urea group or a group that can be converted to a carbamate or urea group
and an isocyanate group and
20 (B) a compound having at least two groups reactive with isocyanate functionality,
and mixtures thereof;
(b) an acrylic polymer comprising active hydrogen-containing functional groups that are reactive
25 with component (c), and
(c) a curing agent that is reactive with compound (a) and compound (b),

wherein said carbamate group has a structure



in which R is H or alkyl, and further wherein said urea group has a structure



in which R' and R'' are each independently H or alkyl,
10 or R' and R'' together form a heterocyclic ring structure.

2. A composition according to claim 1, wherein component (a) has at least one carbamate group.

15

3. A composition according to claim 1, wherein component (a) comprises compound (a)(1).

4. A composition according to claim 3, wherein
20 component (a)(1) comprises at least one carbamate group.

5. A composition according to claim 3, wherein compound (a)(1)(A) is monomeric.

25

6. A composition according to claim 3, wherein compound (a)(1)(A) has a group selected from the group consisting of carbamate groups, urea groups, and combinations thereof.

30

7. A composition according to claim 3, wherein compound (a)(1)(A) has a carbamate group.

8. A composition according to claim 3, wherein compound (a)(1)(A) is a hydroxyalkyl carbamate.
- 5 9. A composition according to claim 3, wherein compound (a)(1)(A) is a hydroxyalkyl cyclic carbonate.
10. A composition according to claim 3, wherein the group on compound (a)(1)(A) that is reactive with
10 (a)(1)(B) is hydroxyl or amino.
11. A composition according to claim 3, wherein compound (a)(1)(A) has one group that is reactive with (a)(1)(B).
- 15 12. A composition according to claim 3, wherein compound (a)(1)(B) is monomeric.
13. A composition according to claim 3, wherein
20 compound (a)(1)(B) is an isocyanurate.
14. A composition according to claim 3, wherein compound (a)(1)(B) is a monomeric, isocyanate-functional reaction product of a diisocyanate and a
25 polyol.
15. A composition according to claim 1, wherein component (a) comprises compound (a)(2).

16. A composition according to claim 15, wherein compound (a)(2)(A) has, on average, one isocyanate group per molecule.
- 5 17. A composition according to claim 15, wherein compound (a)(2)(A) has a carbamate or urea group.
18. A composition according to claim 15, wherein compound (a)(2)(A) has a carbamate group.
- 10 19. A composition according to claim 15, wherein compound (a)(2)(A) is a reaction product of an hydroxyalkyl carbamate and a monomeric isocyanate compound.
- 15 20. A composition according to claim 15, wherein compound (a)(2)(A) is a reaction product of an hydroxyalkyl cyclic carbonate and a monomeric isocyanate compound.
- 20 21. A composition according to claim 15, wherein compound (a)(2)(B) is a diamine.
22. A composition according to claim 15, wherein
25 compound (a)(2)(B) is a polyol.
23. A composition according to claim 15, wherein compound (a)(2)(B) is a diol.
- 30 24. A composition according to claim 1, wherein component (b) has functionality selected from the group

consisting of carbamate functionality, urea functionality, hydroxyl functionality, and combinations thereof.

- 5 25. A composition according to claim 1, wherein component (b) has functionality selected from the group consisting of carbamate functionality, hydroxyl functionality, and combinations thereof.
- 10 26. A composition according to claim 1, wherein component (b) has carbamate functionality.
27. A composition according to claim 1, wherein component (b) has hydroxyl functionality.
- 15 28. A composition according to claim 1, wherein component (c) is an aminoplast.
29. A composition according to claim 1, wherein
- 20 component (c) is a melamine formaldehyde resin.
30. A composition according to claim 1, wherein component (c) is a urea formaldehyde resin.
- 25 31. A composition according to claim 1, wherein R and R'' are each independently H or alkyl of from 1 to about 4 carbon atoms.
32. A composition according to claim 1, wherein R and
- 30 R'' are each H.

33. A composition according to claim 1, wherein the composition is a clearcoat coating composition.

34. A composition according to claim 1, further
5 comprising a pigment.

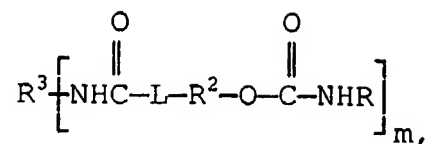
35. An article comprising a substrate having thereon a cured coating derived from a coating composition according to claim 1.

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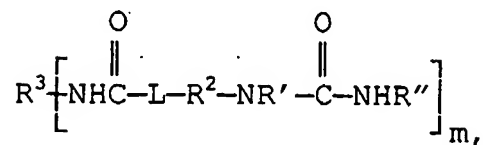
36. An article according to claim 51, wherein said substrate is a flexible substrate.

37. A curable coating composition, comprising
15 (a) a compound selected from the group consisting of compounds having structures

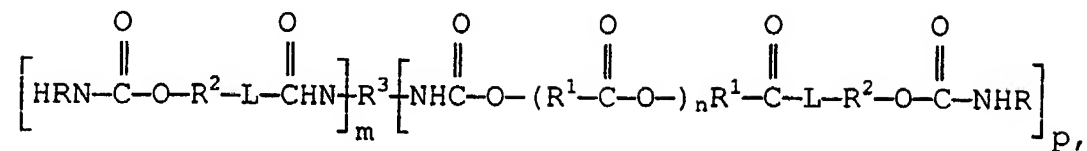
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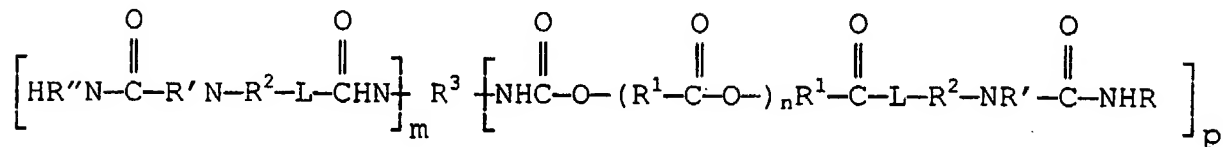
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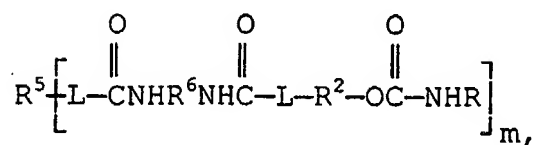
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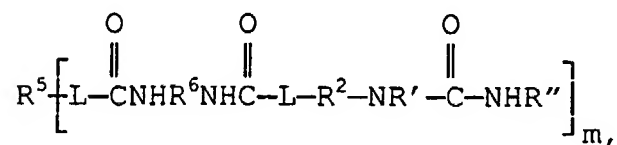
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and mixtures thereof; and

- (b) an acrylic polymer comprising active hydrogen-containing functional groups that are reactive
5 with component (c), and
(c) a curing agent that is reactive with compound (a) and component (b),

wherein R is H or alkyl; R' and R'' are each independently H or alkyl, or R' and R'' together form a
10 heterocyclic ring structure; R² is alkylene or substituted alkylene; R¹, R³, R⁵, and R⁶ are independently alkylene, cycloalkylene, or arylalkylene, or R³, R⁵, and R⁶ are independently arylene, or a structure that includes a cyanuric ring, a urethane
15 group, a urea group, a carbodiimide group, a biuret structure, or an allophonate group; n is from 0 to about 10; m is from 2 to about 6; and L is O, NH, or NR⁴, where R⁴ is an alkyl; in which p is from 1 to 5, and m + p is 2 to 6; in which R⁵ and R⁶ are each
20 independently alkylene, cycloalkylene, alkylarylene, or arylene, or R⁶ is a structure that includes a cyanuric ring, a biuret structure, or an allophonate group.

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US 99/24703

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C09D201/06 C08G18/38

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C09D C08G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P, X	EP 0 915 113 A (BASF CORP) 12 May 1999 (1999-05-12) claims 1,3,6,8,11	1-14, 24-28, 31-36
A	WO 97 26304 A (PPG INDUSTRIES INC) 24 July 1997 (1997-07-24) claim 1	1

☐ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents:

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Date of the actual completion of the international search

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INTERNATIONAL SEARCH REPORT

information on patent family members

International Application No

PCT/US 99/24703

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